Components

- A. Protocol F instructions for total cyanide analysis in bamboo shoots.
- B. Thirty (30) flat-bottomed plastic bottles with screw capped lids.
- C. Two (2) graduated 1 ml plastic pipettes.
- D. One hundred (100) buffer papers.
- E. One hundred (100) yellow **indicator** papers glued to strips of clear plastic. STORE IN FREEZER. Stable for one month <u>only</u> at room temperature.
- F. Colour chart with ten (10) shades of colour which correspond to 0-800 ppm total cyanide.
- G. Ten (10) pink standard papers with linamarin (ppm cyanide on label). STORE IN REFRIGERATOR.
- H. Ten (10) buffer/enzyme papers, identified by a small black spot.

Method (Complete steps 2 to 6 quickly as the enzyme acts rapidly to release HCN)

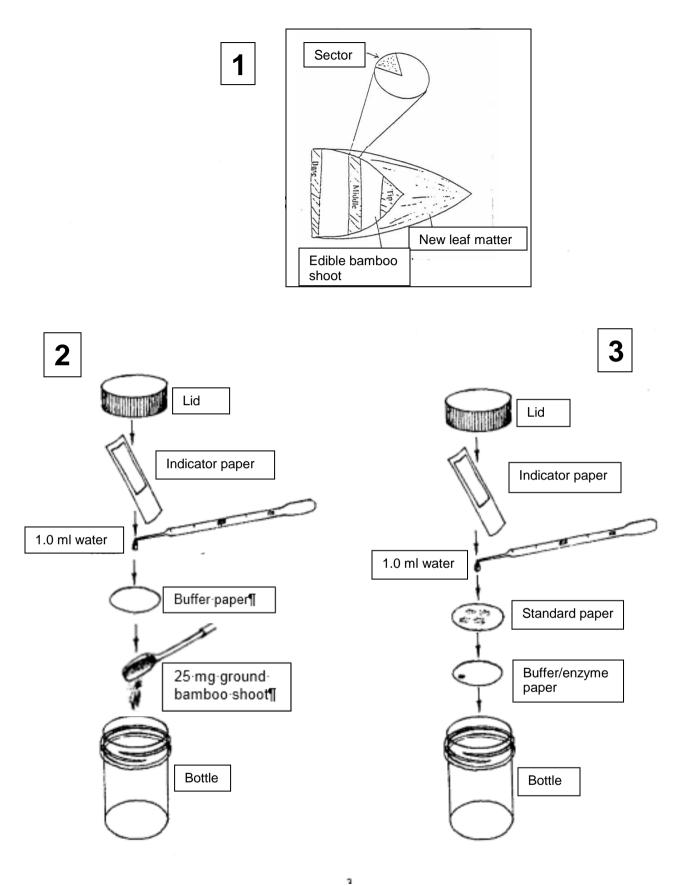
- 1. Follow sketch 1. Remove the new leaf matter to expose the edible bamboo shoot. The tip contains more cyanide than the middle which contains more than the base. *The enzyme in the shoot acts rapidly to produce HCN so complete steps 2 to 6 quickly.* Take a 0.5-1 cm section across the shoot, cut a sector and IMMEDIATELY grind up using a pestle and mortar.
- Follow sketch 2. Accurately weigh about 25mg ground shoot into a flat-bottomed plastic bottle. Weight of sample = Z mg.
- 3. IMMEDIATELY add a buffer paper disc.
- 4. Add 1.0 ml of clean water using the plastic pipette and mix gently.
- 5. IMMEDIATELY add a yellow **indicator** paper attached to a plastic strip so that the paper does not touch the liquid in the bottle. When not in use STORE INDICATOR PAPERS IN FREEZER.
- 6. IMMEDIATELY close the bottle with a screw capped lid.
- 7. A positive and negative control should be run for each set of experiments.
 - a. For a negative control, prepare another sample as shown in sketch 2 but with no bamboo shoot present.
 - b. For a positive control, follow sketch 3. Place a **buffer/enzyme** paper disc in the bottle. Add a pink **standard** paper disc and then 1.0 ml water from a pipette and the yellow **indicator** paper. IMMEDIATELY close the bottle with a screw capped lid.
- 8. Allow the bottles to stand for 16-24 hour at room temperature.
- 9. Open the bottles and match the colour of the **indicator** papers against the shades of colour on the colour chart supplied.
- 10. The total cyanide content in ppm is obtained by multiplying the value from the colour chart by 100/Z. Check that the negative control is zero and the positive control gives a colour equivalent to ppm on label.

THIS SECTION TO BE FOLLOWED IF YOU HAVE A SPECTROPHOTOMETER

- 11. For each sample, carefully remove the plastic backing sheet from the **indicator** paper.
- 12. Place the paper in a test tube and add 5.0 ml of water measured accurately.
- 13. Leave the test tube at room temperature for about 30 min with occasional gentle stirring.
- 14. Measure the absorbance at 510 nm of the solution, subtract the value of the negative control.
- 15. The total cyanide content in ppm is calculated by the equation¹ total cyanide content (ppm) = 396 x absorbance x 100/Z
- 16. The total cyanide content obtained for the same sample of bamboo shoot, from both measurements 10 and 15 should be about the same. Also check the **standard** value agrees using both methods.

Troubleshooting

- The total cyanide content of the pink **standard** paper should be ±10 ppm of that shown on label. If it is not then it is likely something is wrong with the **standard** paper.
- If the **indicator** paper is left at room temperature it gradually becomes darker and after more than one month its colour will be around 10 ppm on the colour chart.
- If the **indicator** paper has been left in bright sunlight it becomes bleached on one side and is no good.
- If you use a bottle which is not gas tight (e.g. the screw cap is cracked) then gas could escape and this would give a low result.



Reference

¹Bradbury, M. G., Egan, S. V. and Bradbury, J. H. (1999) Determination of all forms of cyanogens in cassava roots and cassava products using picrate paper kits. J. Sci. Food Agric., 79, 593-601. Haque, M. R. and Bradbury, J. H. (2002) Total cyanide determination of plants and foods using the picrate and acid hydrolysis methods. Food Chem., 77, 107-114.

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